Name Hour	
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Chemistry Daily Journal Unit #__2__

REMINDER: CHECKPOINT IS AT PROGRESS REPORTS...not being done with this chapter by then will lower your grade

Today's Date	What did I accomplish yesterday?	What are my goals today? What sections, activities, labs do I want to get done today?

Name	Hour
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Chapter 2: Antacids

Essential questions:

- How is science a human endeavor?
- How do scientists work to gather, analyze, communicate and validate data to form and change models?
- How does the structure of a compound determine its properties?
- How does matter undergo changes and how do we use chemical equations?
- How are mixtures different from pure substances?
- What are the characteristics of acids and bases?

Section	Objectives	Learning Resources	Suggested	Date
		(Check those that you complete)	completion date	completed
2-1	Define and distinguish between pure substance (elements and compounds) and mixtures (homogeneous and heterogeneous)	□ Reading guide □ PowerPoint □ Practice 1-1 □ Discuss with peers □ Discuss with teacher		
2-2	Explain and practice nomenclature rules in naming chemicals: binary ionic, multi-valent ionic, polyatomic ionic and covalent	□ Reading Guide □ PowerPoint □ Worksheet 2-2 □ Practice 2-2 □ Discuss with peers □ Discuss with teacher		
2-3	Explain and practice nomenclature rules for writing chemical formulas: binary ionic, multi- valent ionic, polyatomic ionic and covalent	□ Reading Guide □ PowerPoint □ Worksheet 2-3 □ Practice 2-3 □ Discuss with peers □ Discuss with teacher		
2-4	 Define and distinguish between acids and bases Explain and practice nomenclature rules for naming and writing acids and bases 	□ Reading Guide □ PowerPoint □ Worksheet 2-4 □ Practice 2-4 □ Discuss with peers □ Discuss with teacher		

Name	H	our
2-5	 Investigate characteristics of acids and bases Distinguish between strong and weak acids/bases Distinguish between concentration and strong Explain the pH scale and use of indicators 	□ Lab 2-5 (Must complete first) □ Reading Guide □ PowerPoint □ Discuss with peers □ Discuss with teacher
2-6	 Define components of a chemical reaction Practice converting from word equations to formula equations Define double replacement and neutralization reactions 	□ Reading Guide □ PowerPoint 2.6 □ Discuss with peers □ Discuss with teacher
2-7	Use the Law of Conservation of mass to explain need for balanced equations	□ Lab 2-7 (Must complete first) □ Reading Guide □ PowerPoint □ Practice 2-7 □ Worksheet 2-7 □ Discuss with peers □ Discuss with teacher
2-8	 Explain Collision Theory and relate theory to variables that affect the rate of a reaction Define and explain catalysts role in a reaction 	□ Lab 2.8 (Must complete first) □ Reading Guide □ PowerPoint □ Discuss with peers □ Discuss with teacher
Final Chpt 2 Lab	Acid/base reactions and use of indicators.Evaluating household products.	□ Lab (Must complete)
Review & Test	Demonstrate knowledge from Chapter 2	☐ Review section quizzes ☐ Chapter 1 review ☐ Discuss with peers ☐ Discuss with teacher

REMINDER: Checkpoint is at the end of the quarter...not being done with this chapter by then will lower your grade